



**A New Concept Toward Asymmetric Synthesis  
—Chiral relay in auxiliary and catalyst design**

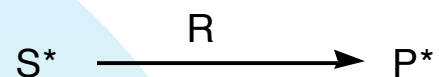
Gang Hu

Department of Chemistry  
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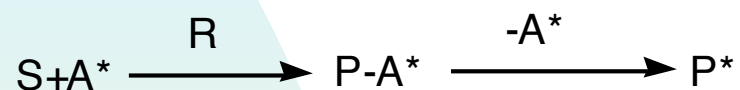
Oct 2., 2002

# Conventional Strategies for Asymmetric Induction

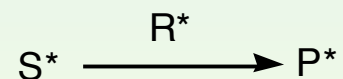
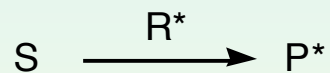
## 1. Substrate Control



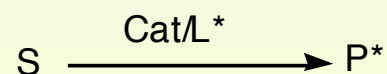
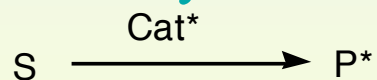
## 2. Auxiliary control



## 3. Reagent control



## 4. Catalyst control

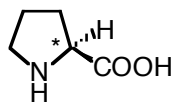


Ling, G.; Cheng, Y.; Cheng, X.; Li, Y. *Asymmetric synthesis---Enantioselective reactions and their applications*. Science Publishing, Beijing, 2000

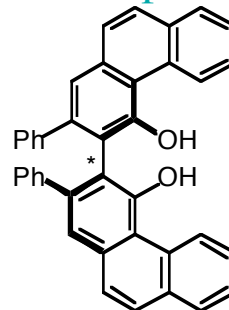
# Can achiral auxiliary or ligand induce asymmetry?

## Conventional Model:

- (1) Rigid, conformationally constricted auxiliary or ligand is usually required
- (2) Stereogenic center would better be close to prochiral reactive site

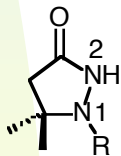


L-Proline

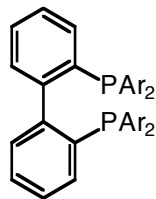


(R)-VAPOL

## Conformationally flexible species:



Pyrrolidinone

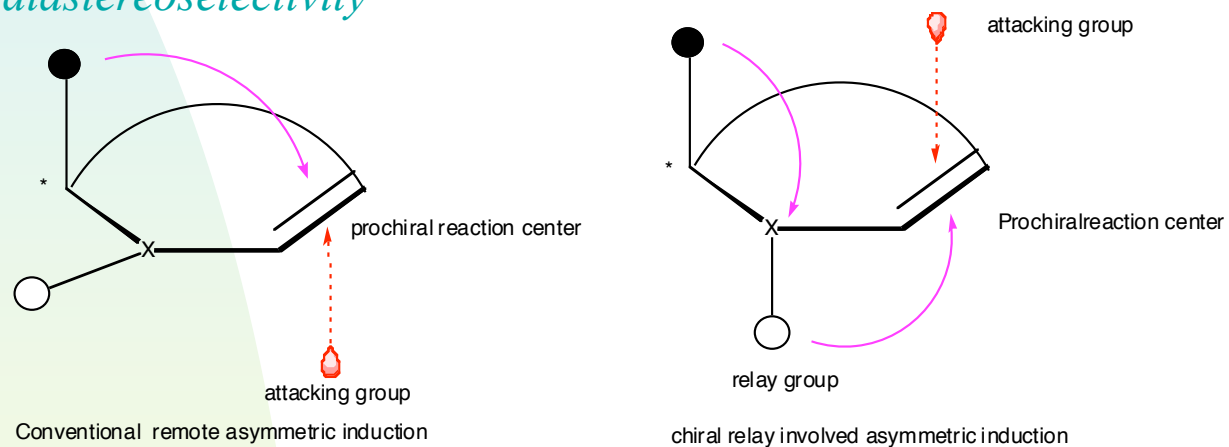


BIPHEP

# The Concept of Chiral Relay

First explicitly proposed by Davies in 1998 as follows:

*“an achiral conformationally flexible group is inserted between the stereogenic center and the prochiral reaction center.....In ideal circumstances, the conformationally flexible group should serve to both relay, and amplify the stereochemical information of the existing stereogenic center, thus enabling efficient control of diastereoselectivity”*



**\*A group of sloppy lazy men are trained into aggressive Spartans with iron will by a strict commander.**

# Two principal kinds of chiral relay

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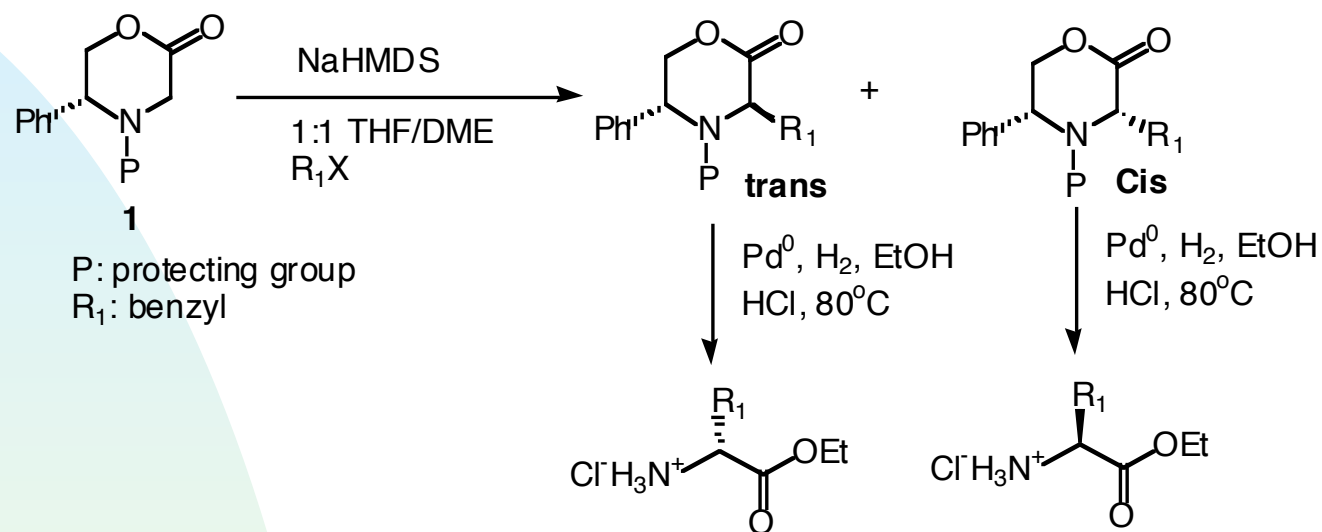
## 1. Chiral relay in auxiliary

- a. Stereogenic center fixed on the auxiliary
- b. Lewis acid mediated chiral relay

## 2. Chiral relay in catalyst

- a. Cooperative functions of flexible ligand and chiral ligand
- b. Independent function of flexible ligand

## Chiral Relay in Auxiliary

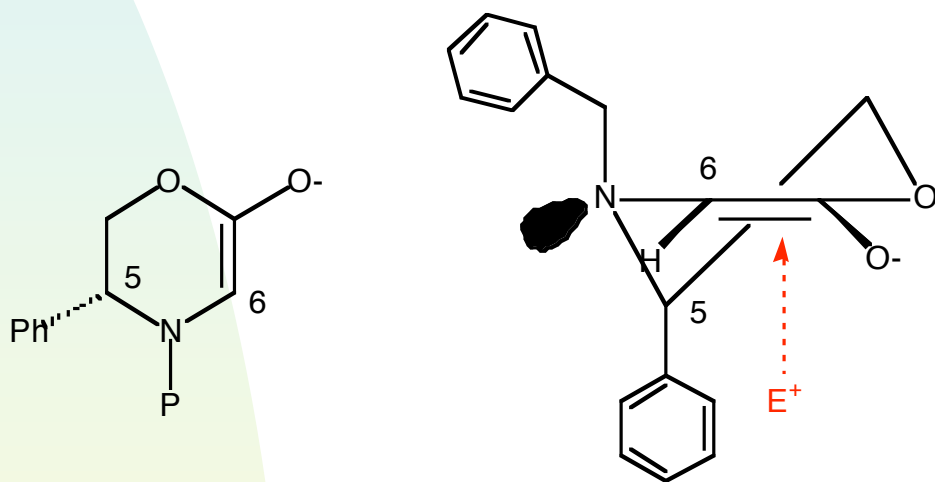


Entry	Protecting group	X	Trans:Cis	Yield
1	Boc (-OCO- <sup>t</sup> Bu)	Br	200:1	78%
2	Cbz (-COOCH <sub>2</sub> Ph)	I	200:1	85%
3	Bn (-CH <sub>2</sub> Ph)	Br	1:17	83%
4	Me	Br	1:18	

## Chiral Relay in Auxiliary

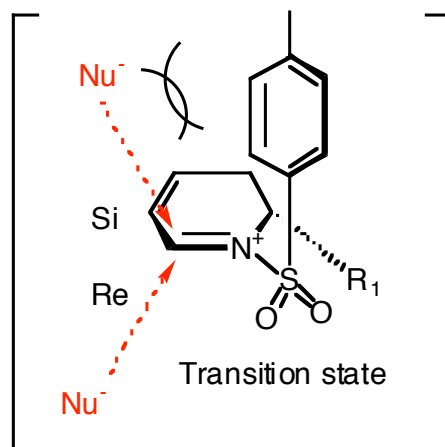
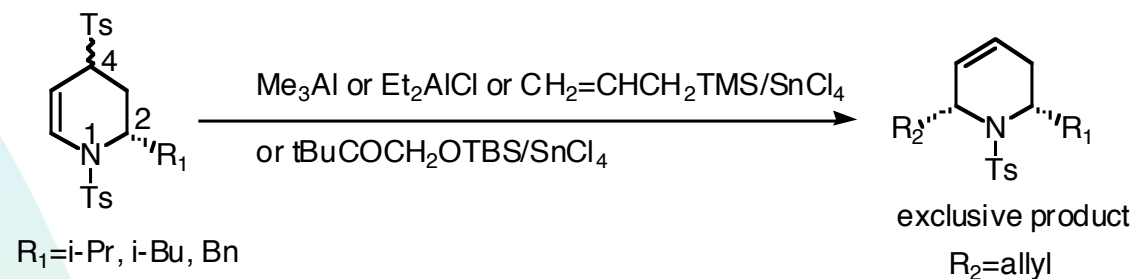
Effects being considered for the conformational preference:

1. Electronic effect
2. Steric effect



preferred half-chair conformation of the enolate

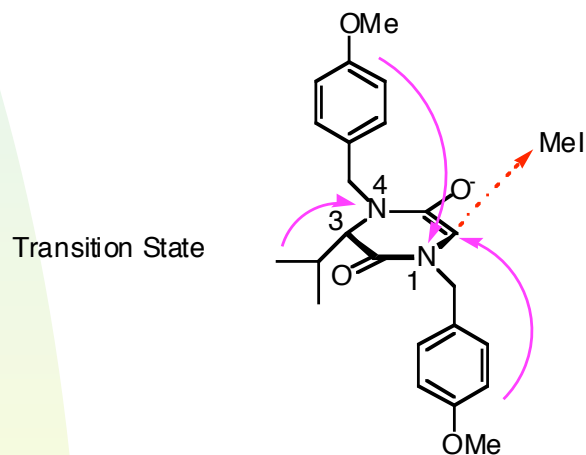
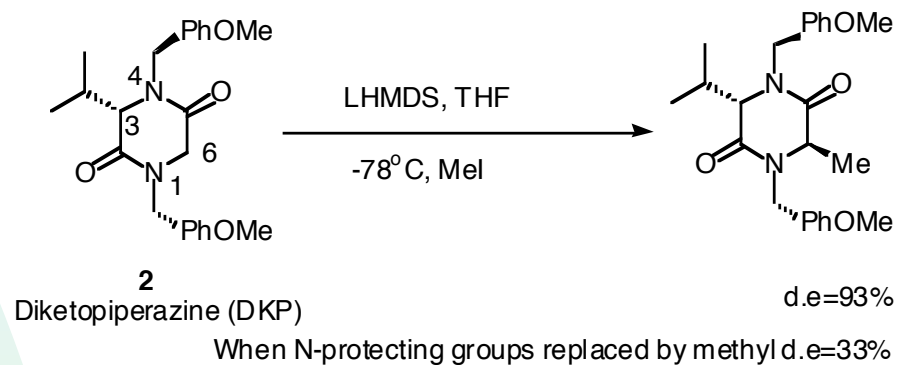
# Chiral Relay in Auxiliary





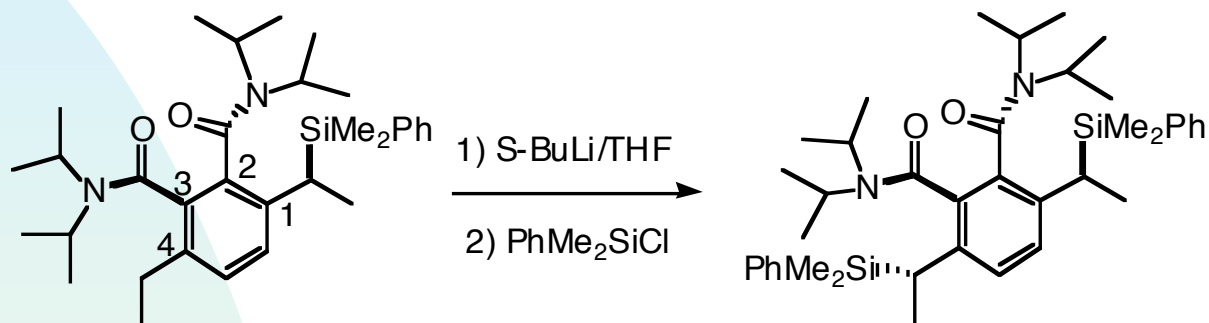
# Chiral Relay in Auxiliary

## Relay network

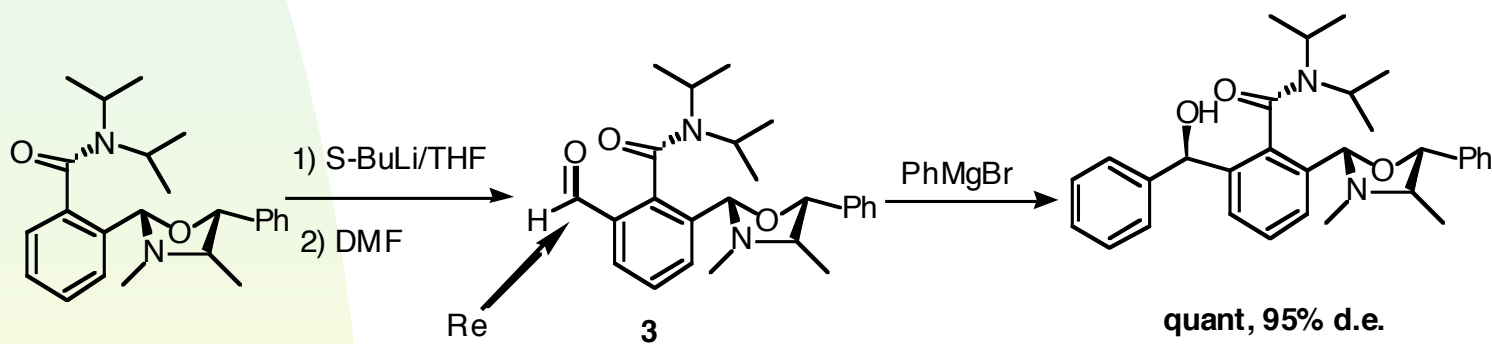


# Chiral Relay in Auxiliary

## Tertiary amide as relay group



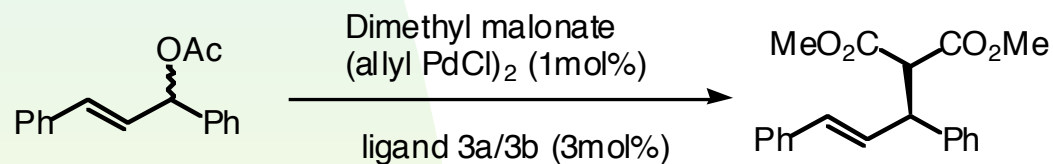
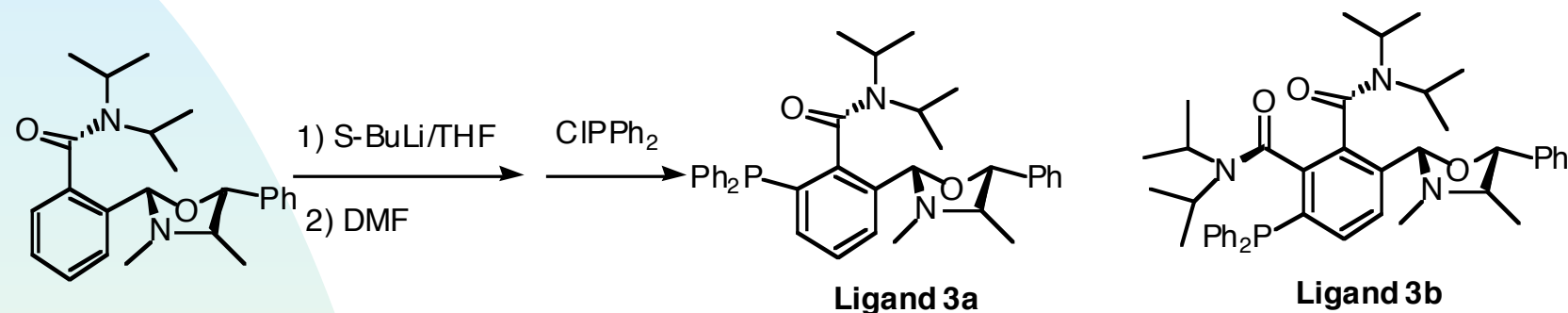
77% yield, only one diastereomer



Clayden, J.; Pink, J. H.; Yasin, S. A. *Tet. Lett.* **1998**, 39, 105

Clayden, J.; Lai, L. W.; Helliwell, M. *Tet. Asymmetry* **2001**, 12, 695-698

## Chiral Relay in Auxiliary



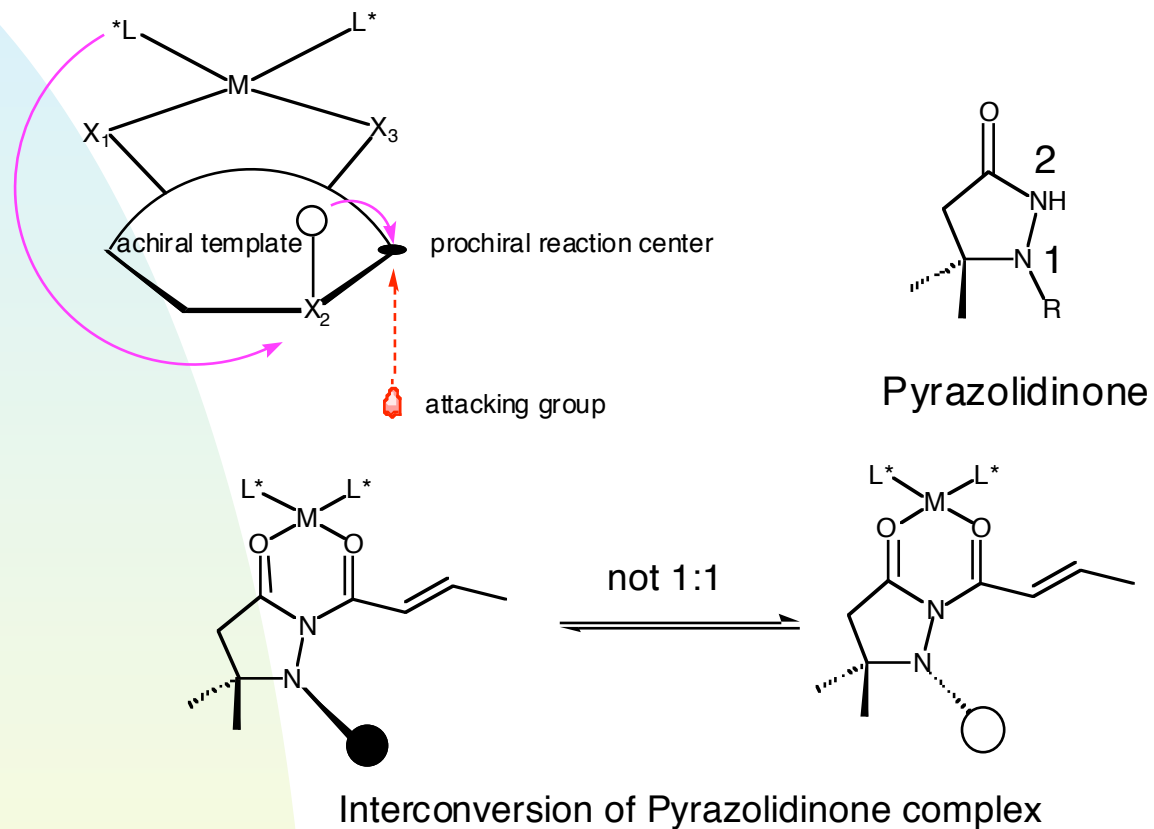
Ligand 3a: (S)-(-), 82%ee, 93% yield  
Ligand 3b: (R)-(+), 53%ee, 85% yield

Clayden, J.; Lai, L. W.; Helliwell, M. *Tet. Asymmetry* **2001**, 12, 695-698

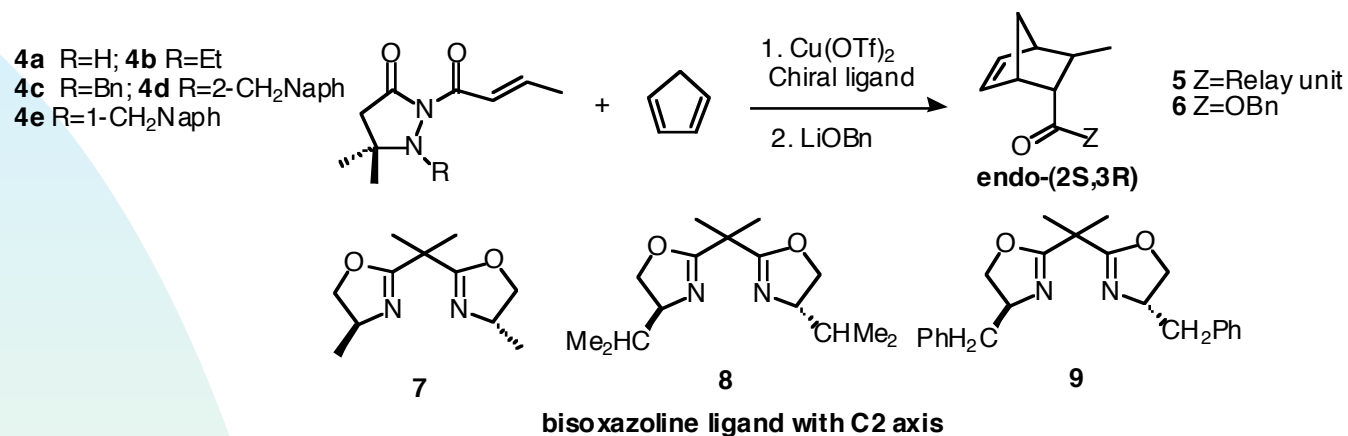
For review of asymmetric allylic alkylation, see Trost, B. M. *Chem. Rev.* **1996**, 96, 395-422

# Lewis acid mediated chiral relay

## Chiral relay controlled by labile group

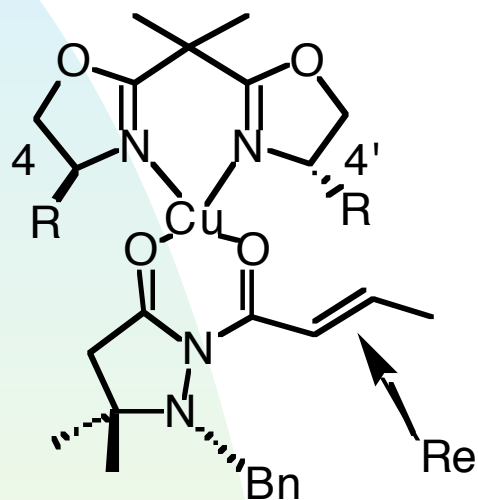


## Lewis acid mediated chiral relay

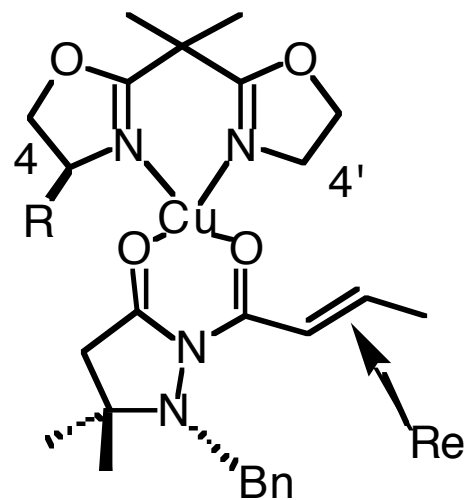


Run	Substrate	7 ee(endo)	8 ee(endo)	9 ee(endo)
1	4a R=H	29%	08%	03%
2	4b R=Et	64%	56%	55%
3	4c R=Bn	71%	71%	71%
4	4d R=2-CH <sub>2</sub> Naph	79%	65%	69%
5	4e R=1-CH <sub>2</sub> Naph	86%	92%	85%

## Lewis acid mediated chiral relay

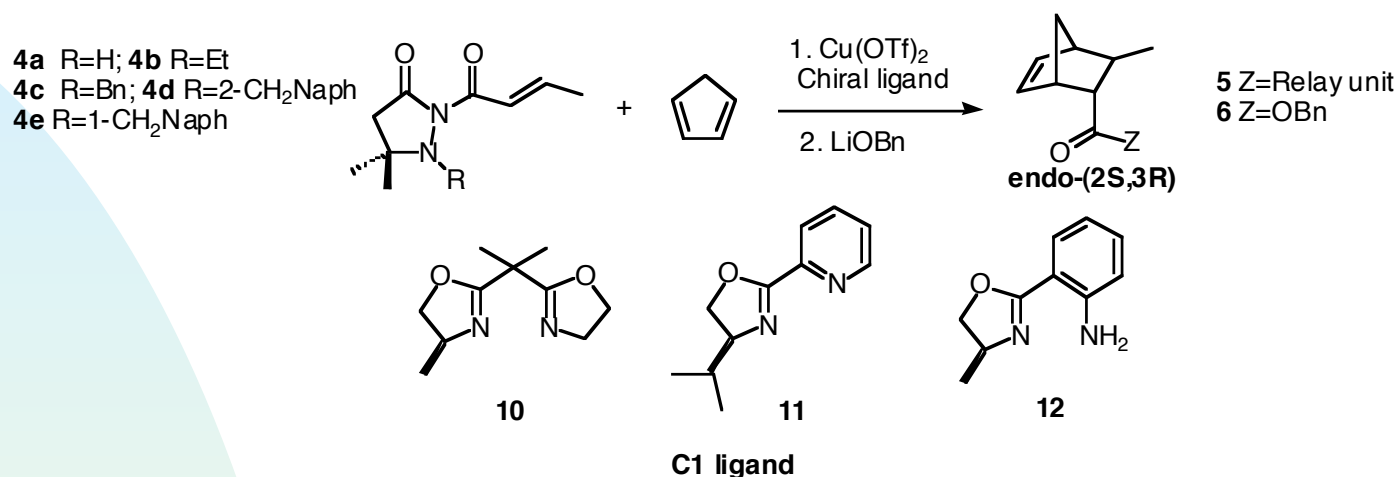


Complex with C2 ligand



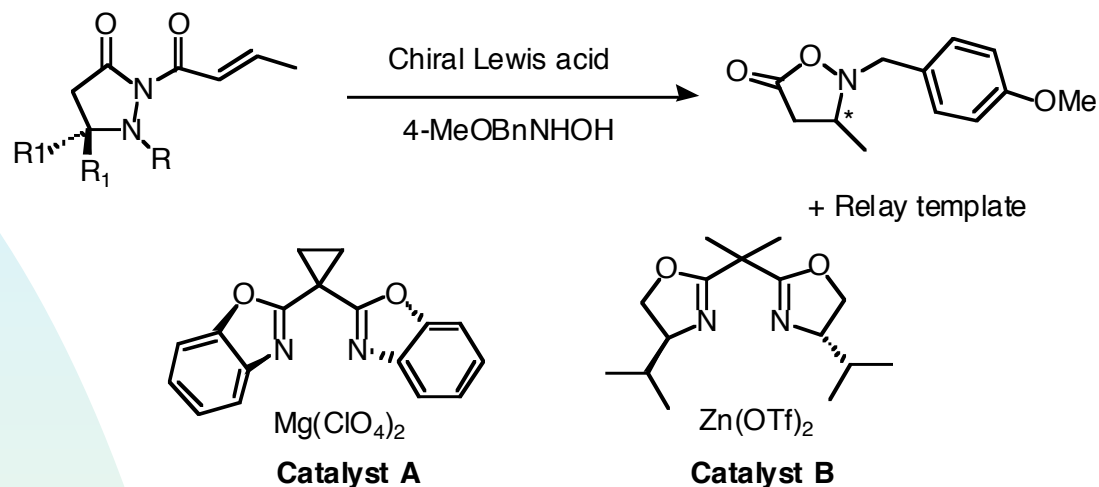
Complex with C1 ligand

# Lewis acid mediated chiral relay



Run	Substrate	10 <i>ee(endo)</i>	11 <i>ee(endo)</i>	12 <i>ee(endo)</i>
1	<b>4a</b> R=H	04%	01%	01%
2	<b>4b</b> R=Et	29%	12%	26%
3	<b>4c</b> R=Bn	47%	38%	51%
4	<b>4d</b> R=2-CH <sub>2</sub> Naph	56%	59%	58%
5	<b>4e</b> R=1-CH <sub>2</sub> Naph	69%	66%	71%

# Lewis acid mediated chiral relay



run	Substrate	Catalyst A		Catalyst B	
		Yield	ee%	Yield	ee%
1	R <sub>1</sub> =Me, R=H	75%	76(R)*	74%	28(S)
2	R <sub>1</sub> =Me, R=Et	70%	52(R)	74%	49(S)
3	R <sub>1</sub> =Me, R=Bn	67%	78(R)	71%	68(S)
4	R <sub>1</sub> =Me, R=2-CH <sub>2</sub> Naph	75%	78(R)	73%	70(S)
5	R <sub>1</sub> =Me, R=1-CH <sub>2</sub> Naph	77%	81(R)	75%	75(S)

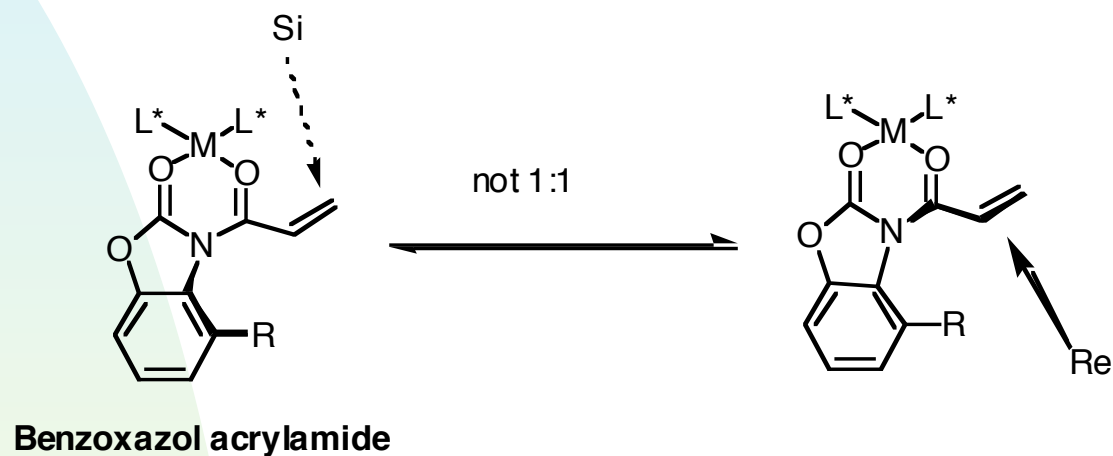
Sibi, M. P.; Liu, M. *Org. Lett.* **2001**, 3, 4181-4184

\* The unusually high ee is presumably attributed to a very different chelation pattern

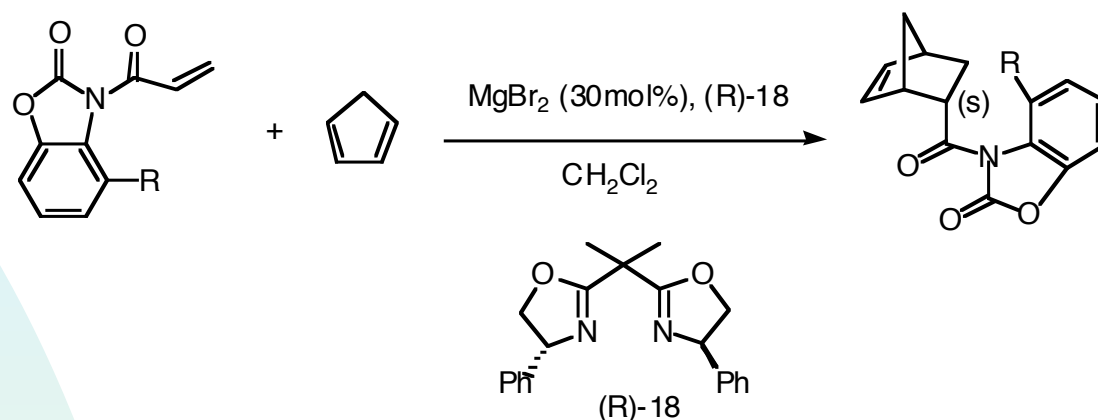


# Lewis acid mediated chiral relay

## Chiral relay controlled by axis



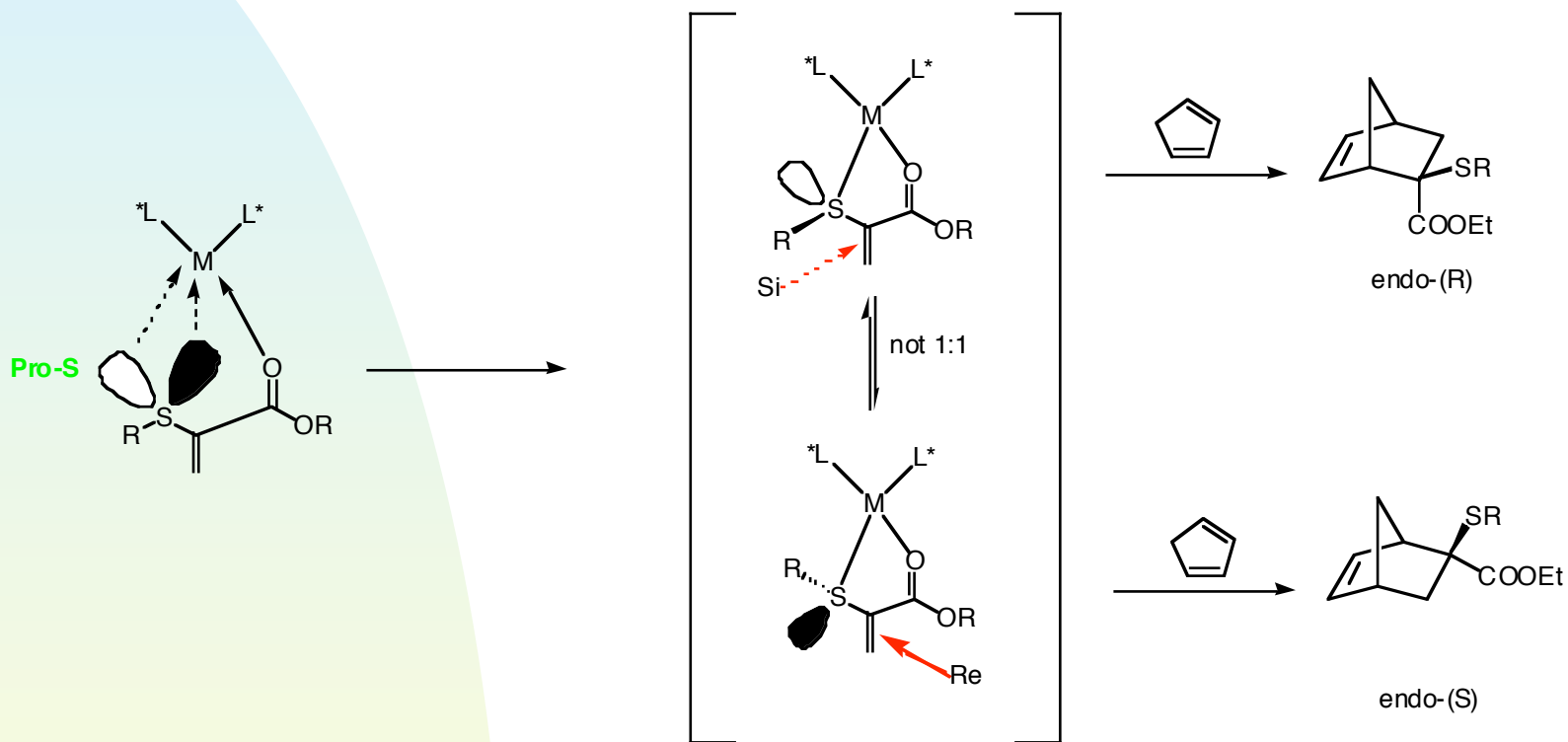
## Lewis acid mediated chiral relay



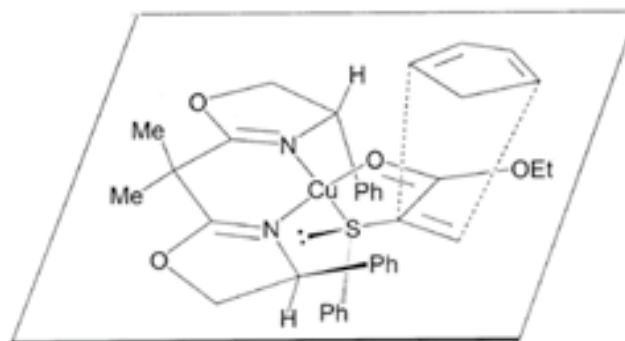
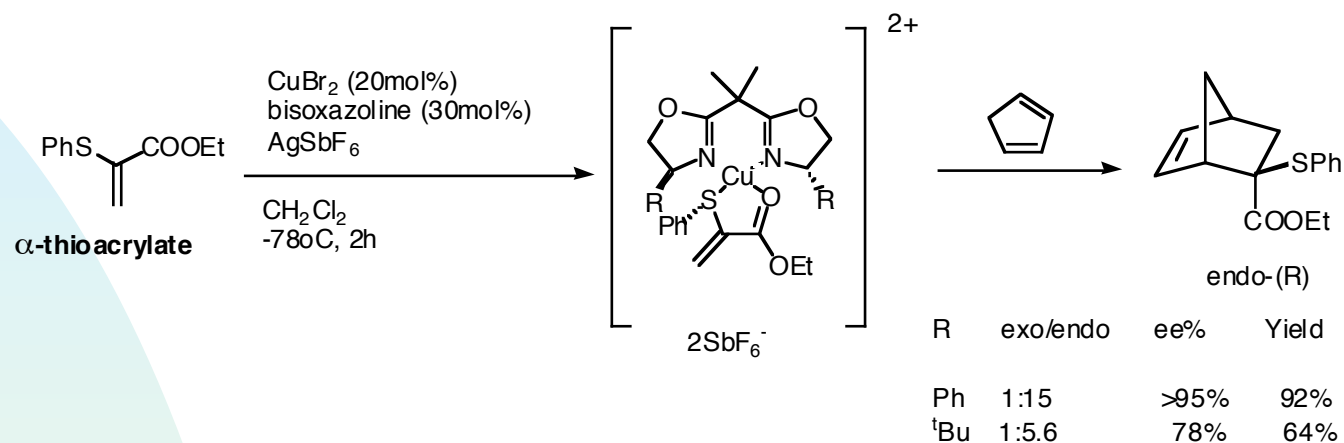
Run	R	Yield (endo/exo)	ee (endo)
1	H	98%(22:1)	14%(s)
2	Me	97%(32:1)	74%(s)
3	Et	97%(33:1)	76%(s)
4	Bn	97%(11:1)	86%(s)
5	PMB	98%(16:1)	88%(s)
6	$\text{CH}_2\text{t-Bu}$	98%(20:1)	10%(s)
7	$\text{CH}(\text{Ph})_2$	97%(17:1)	72%(s)
8	$\text{SiMe}_3$	95%(40:1)	40%(s)

## Lewis acid mediated chiral relay

Chiral relay controlled by selectively binding to lone pair

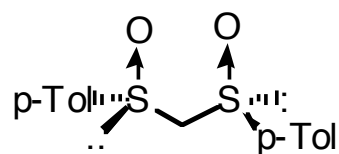
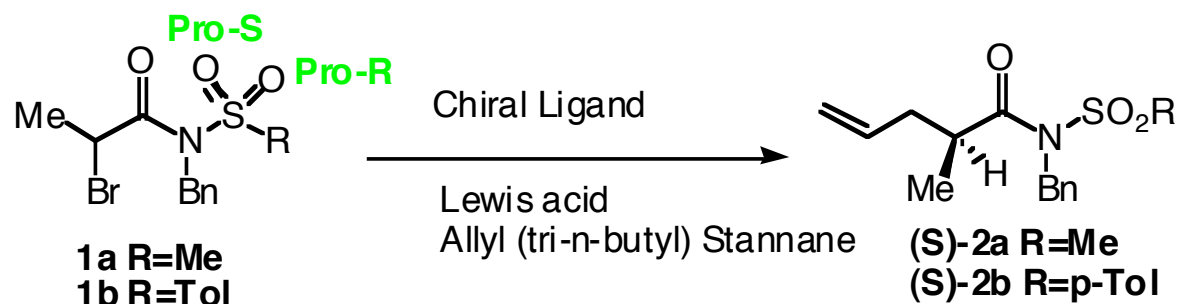


# Lewis acid mediated chiral relay

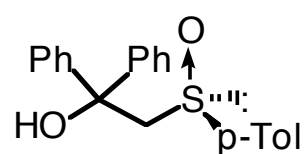


Proposed transition state

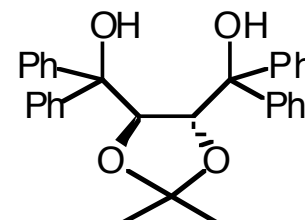
## Lewis acid mediated chiral relay



**3a**



**3b**

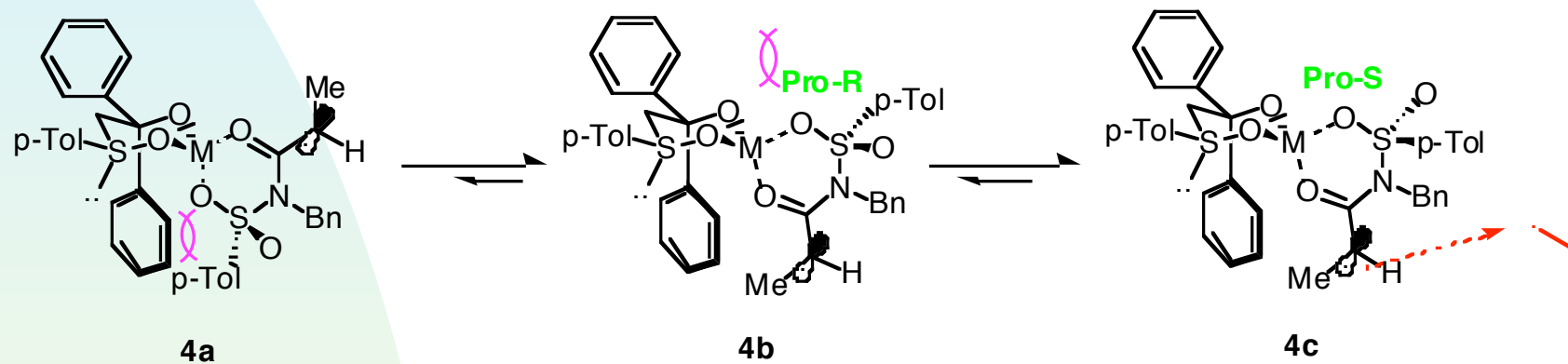


**3c**

Chiral Ligand:

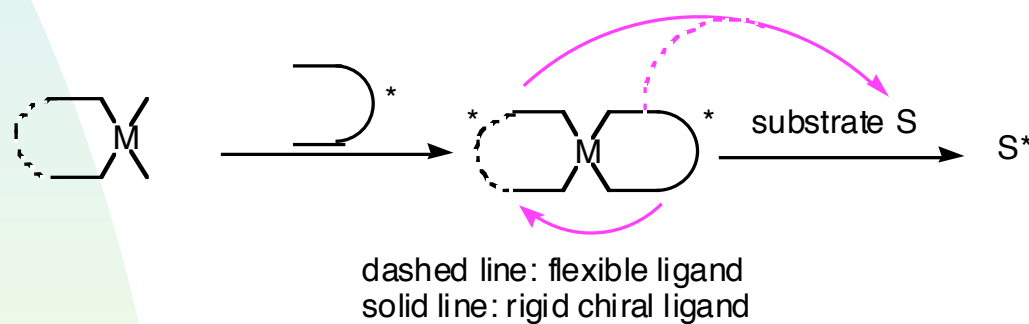
Entry	R	Chiral ligand	Lewis acid	Yield	<i>ee</i>
1	p-Tol	3a	Mg(OTf) <sub>2</sub>	65%	52%
2	p-Tol	3b	Mg(OTf) <sub>2</sub>	41%	83%
3	p-Tol	3c	Ti(O- <i>ipr</i> ) <sub>4</sub>	44%	50%
4	methyl	3c	Ti(O- <i>ipr</i> ) <sub>4</sub>	33%	5%

## Lewis acid mediated chiral relay

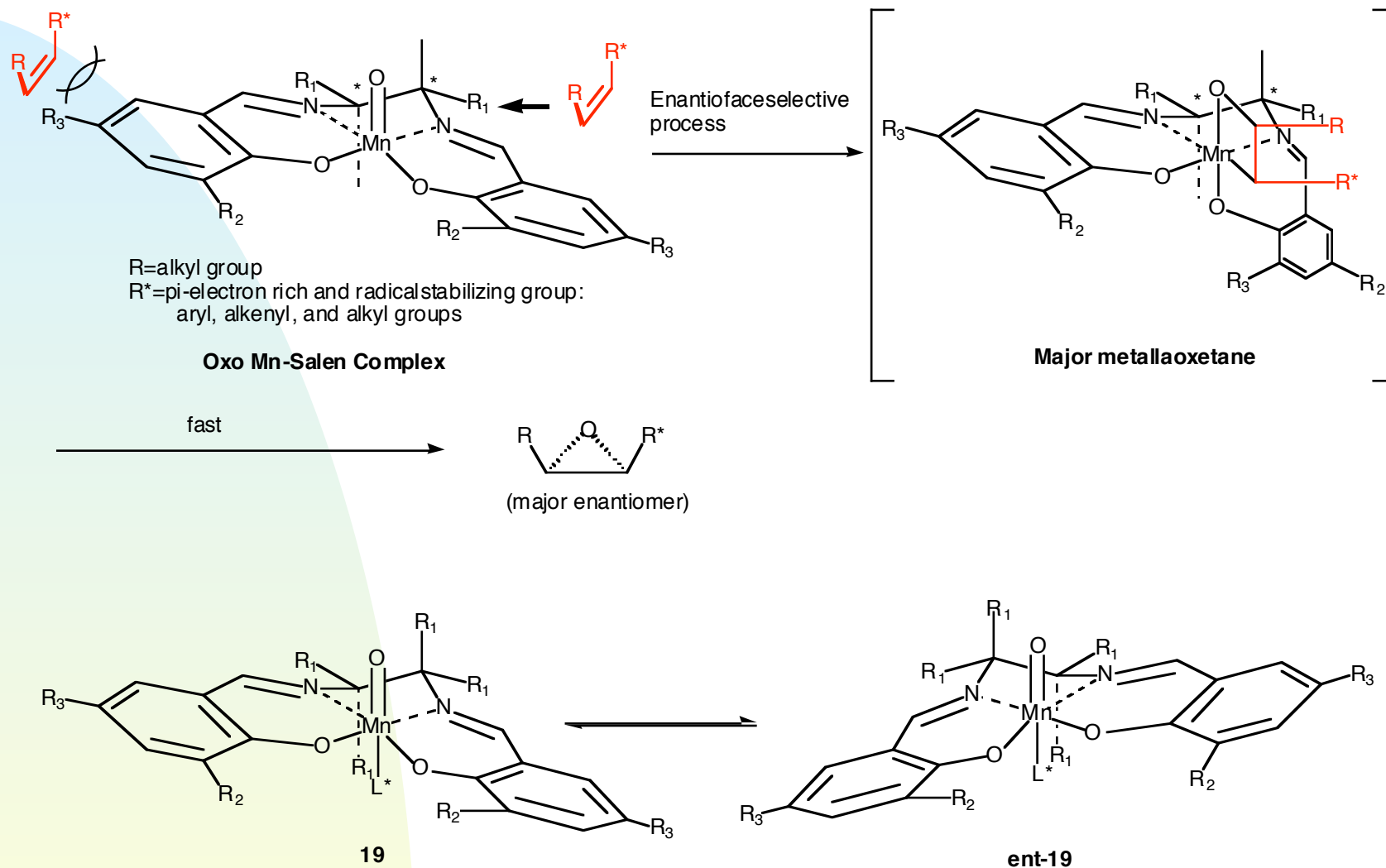


# Chiral Relay in Catalyst

Illustration of the chiral relay in catalyst with the original stereogenic center



# Chiral Relay in Catalyst

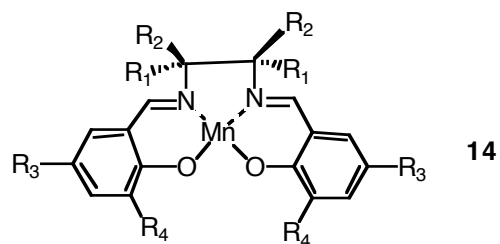
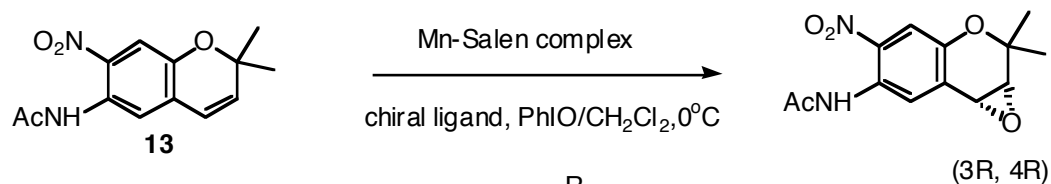


Hashihayata, T; Ito, Y.; Katsuki, T. *Synlett* **1996**, 1079-1081

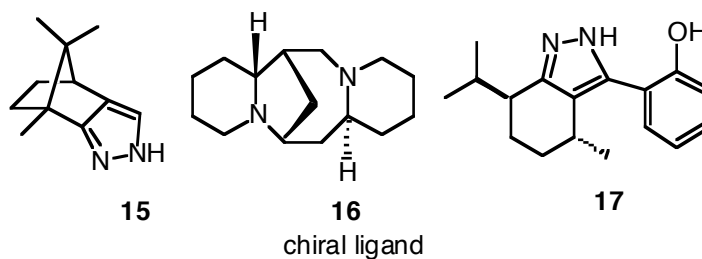
Hashihayata, T.; Ito, Y.; Katsuki, T. *Tetrahedron*, **1997**, 53, 9541



# Chiral Relay in Catalyst

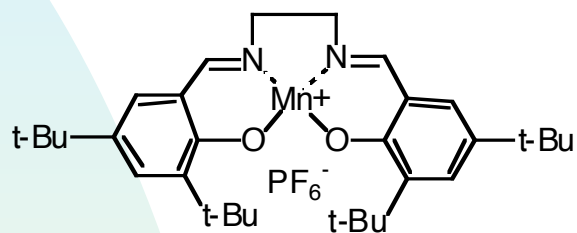
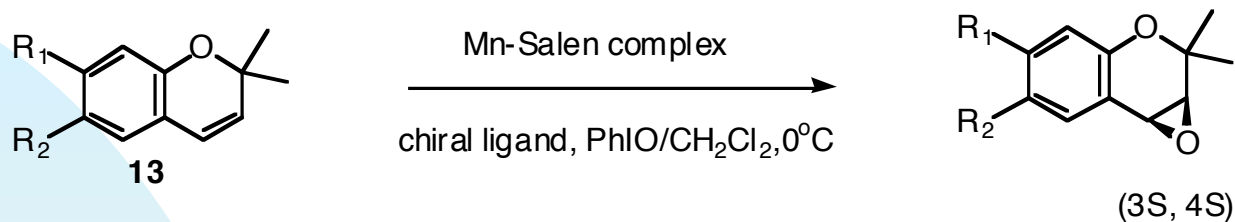


meso Mn-Salen complex

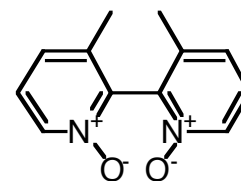


Entry	Mn-Salen complex	Ligand	ee
1	Mn(OAc) <sub>2</sub> or Mn(OAc) <sub>3</sub>	16	0%
2	14a: R <sub>1</sub> =R <sub>2</sub> =R <sub>3</sub> =R <sub>4</sub> =t-Bu, X=PF <sub>6</sub>	15	3%
3	14b: R <sub>1</sub> =R <sub>2</sub> =H, R <sub>3</sub> =R <sub>4</sub> =t-Bu, X=OAc	15	6%
4	14c: n=R <sub>1</sub> =R <sub>2</sub> =H, R <sub>3</sub> =R <sub>4</sub> =t-Bu, X=OAc	16	18%
5	14f: R <sub>1</sub> =R <sub>2</sub> =Me, R <sub>3</sub> =R <sub>4</sub> =t-Bu, X=PF <sub>6</sub>	16	60%
6	14i: R <sub>1</sub> =R <sub>2</sub> =Me, R <sub>3</sub> =OSi(Pr-i) <sub>3</sub> , R <sub>4</sub> =t-Bu, X=PF <sub>6</sub>	16	52%

# Chiral Relay in Catalyst



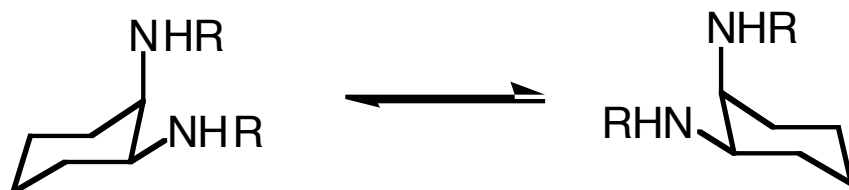
meso Mn-Salen complex



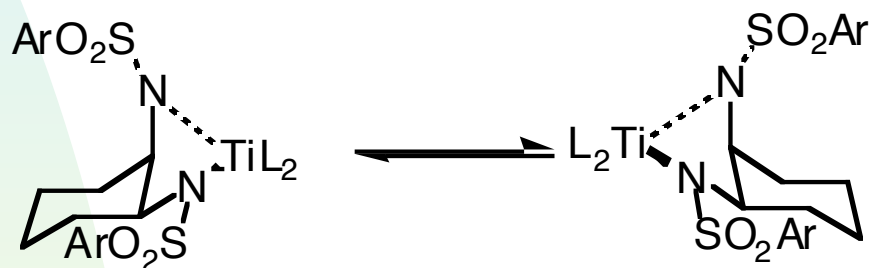
(+)-3,3'-dimethyl-2,2'-bipyridine N,N'-dioxide  
Chiral Ligand

Entry	R <sub>1</sub>	R <sub>2</sub>	Yield	ee	Config.
1	NO <sub>2</sub>	AcHN	65%	82%	3S,4S
2	H	NC	59%	77%	3S,4S
3	=N(O)-O-N=		90%	78%	3S,4S

## Chiral Relay in Catalyst

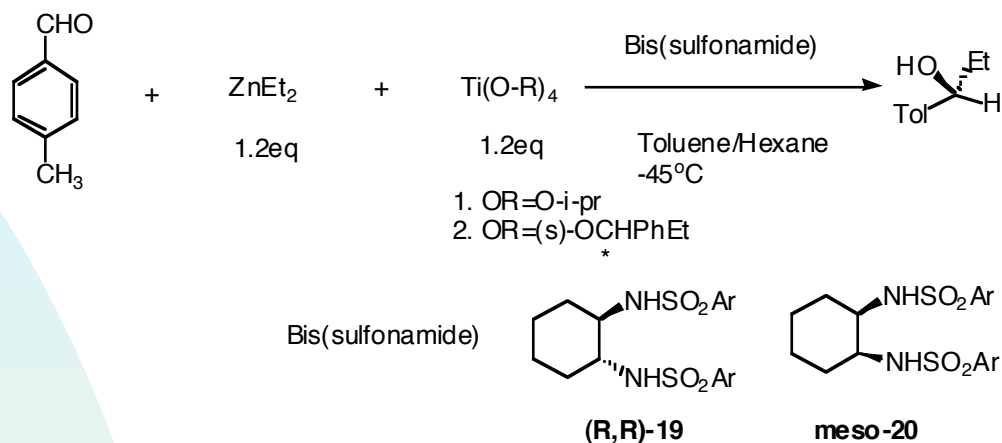


interconversion of diaminocyclohexane



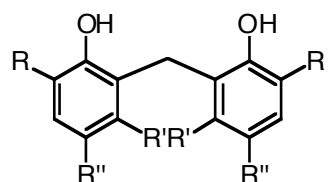
desymmetrization when chelate to chiral LA

# Chiral Relay in Catalyst

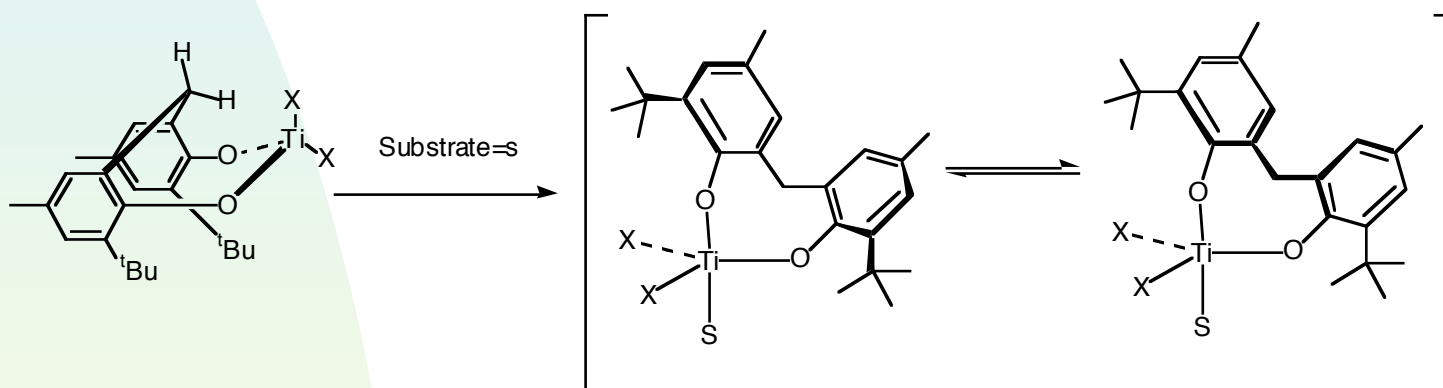


run	R	Ligand	Ar	ee
1	i-pr	(R,R)-19	2,4-C <sub>6</sub> H <sub>3</sub> -Me <sub>2</sub>	79%(S)
2	s-CHPhEt	(R,R)-19	2,4-C <sub>6</sub> H <sub>3</sub> -Me <sub>2</sub>	84%(S)
3	s-CHPhEt	(S,S)-19	2,4-C <sub>6</sub> H <sub>3</sub> -Me <sub>2</sub>	81%(R)
4	s-CHPhEt	---	---	42%(S)
5	s-CHPhEt	20a	4-C <sub>6</sub> H <sub>4</sub> -CMe <sub>3</sub>	84%(R)
6	s-CHPhEt	20b	4-C <sub>6</sub> H <sub>3</sub> -OMe	78%(R)

# Chiral Relay in Catalyst

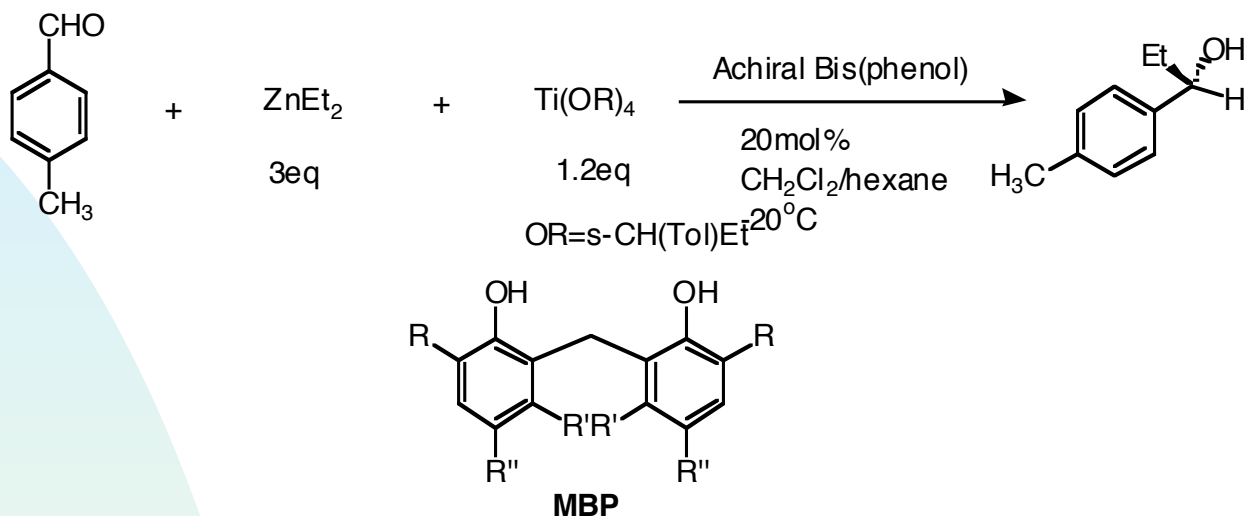


2,2'-methylene-Bisphenol (MBP)



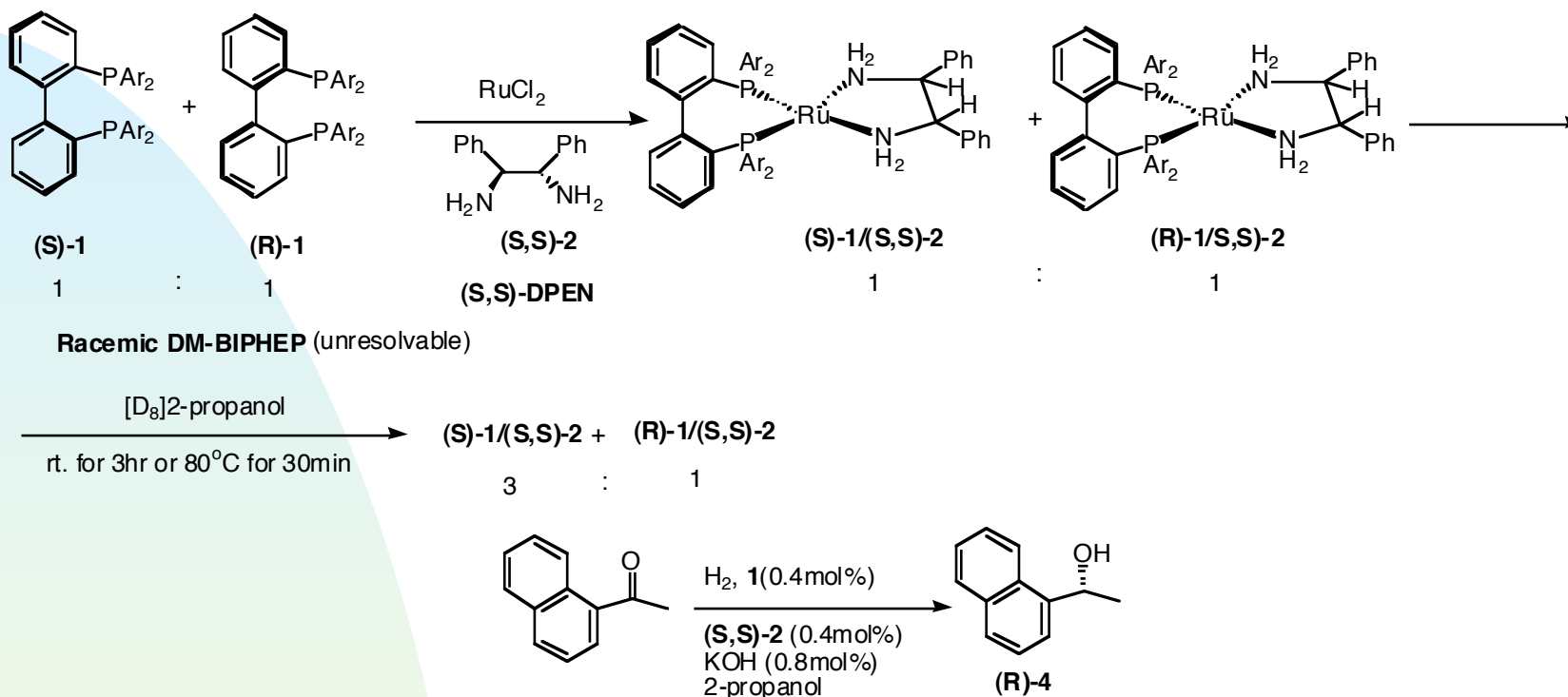
Davis, T. J.; Balsells, J.; Carroll, P. J.; Walsh, P. J. *Org. Lett.* **2001**, 3, 2161

# Chiral Relay in Catalyst



Entry	Ligand	R	R'	R''	ee%
1	No MBP				39(S)
2	1a	H	H	H	1(S)
3	1c	Cl	Cl	Cl	24(S)
4	1d	Me	H	Me	16(S)
5	1e	Ph	H	H	36(S)
6	1f	t-Bu	H	t-Bu	68(S)
7	1g	t-Bu	H	Me	79(S)
8	1h	t-Bu	H	H	73(S)
9	1i	Adamantyl	H	Me	83(S)

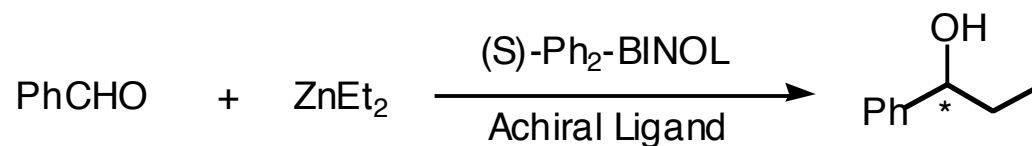
# Chiral Relay in Catalyst

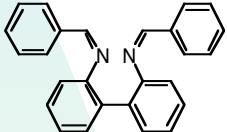
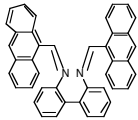
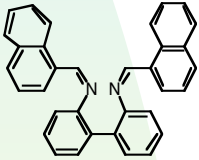
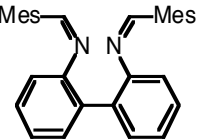
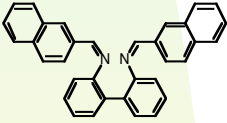
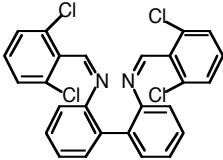


Entry	Phosphane	(S)/(S,S):(R)/(S,S)	ee	Yield
1	DM-BIPHEP	1:1	63%	99%
2	DM-BIPHEP	2:1	73%	99%
3	DM-BIPHEP	3:1	84%	99%
4*	DM-BIPHEP	3:1	92%	99%

Mikami, K.; Korenaga, T.; Terada, M.; Ohkuma, T.; Pham, T.; Noyori, R. *Angew. Chem. Int. Ed.* **1999**, *38*, 495-497 \* The experiment was performed under  $-35^\circ\text{C}$  instead of  $28^\circ\text{C}$  as in other entries

# Chiral Relay in Catalyst

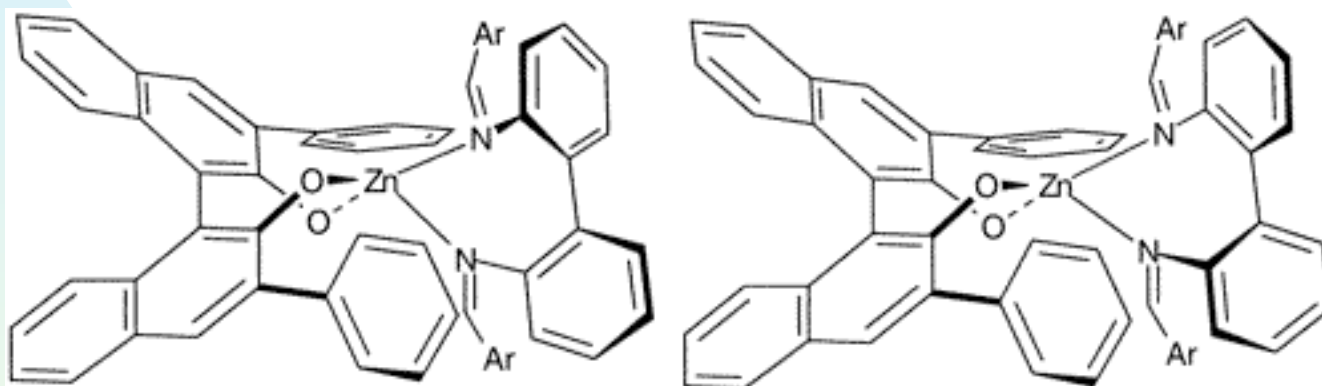


Entry	Achiral Ligand	ee/ 0°C(config.)	Entry	Achiral Ligand	ee/ 0°C(config.)
1		38%(S)	4		54%(R)
2		25%(R)	5		38%(S)
3		27%(S)	6		89%(R)

Costa, A. M.; Jimeno, C.; Gavenonis, J.; Carroll, P. J. Walsh, P. J. *J. Am. Chem. Soc.* **2002**, 124, 6929-6941

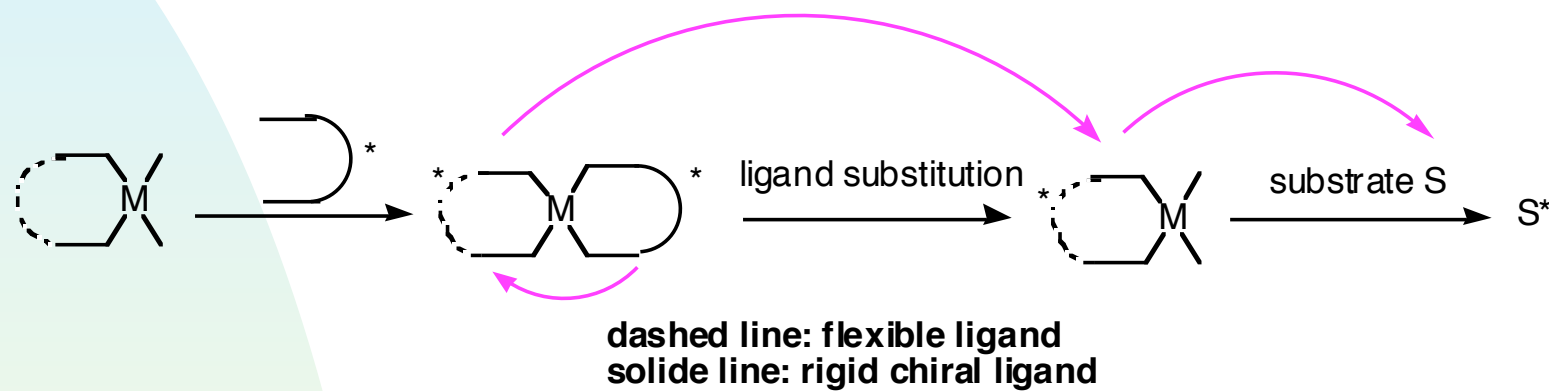


## Chiral Relay in Catalyst

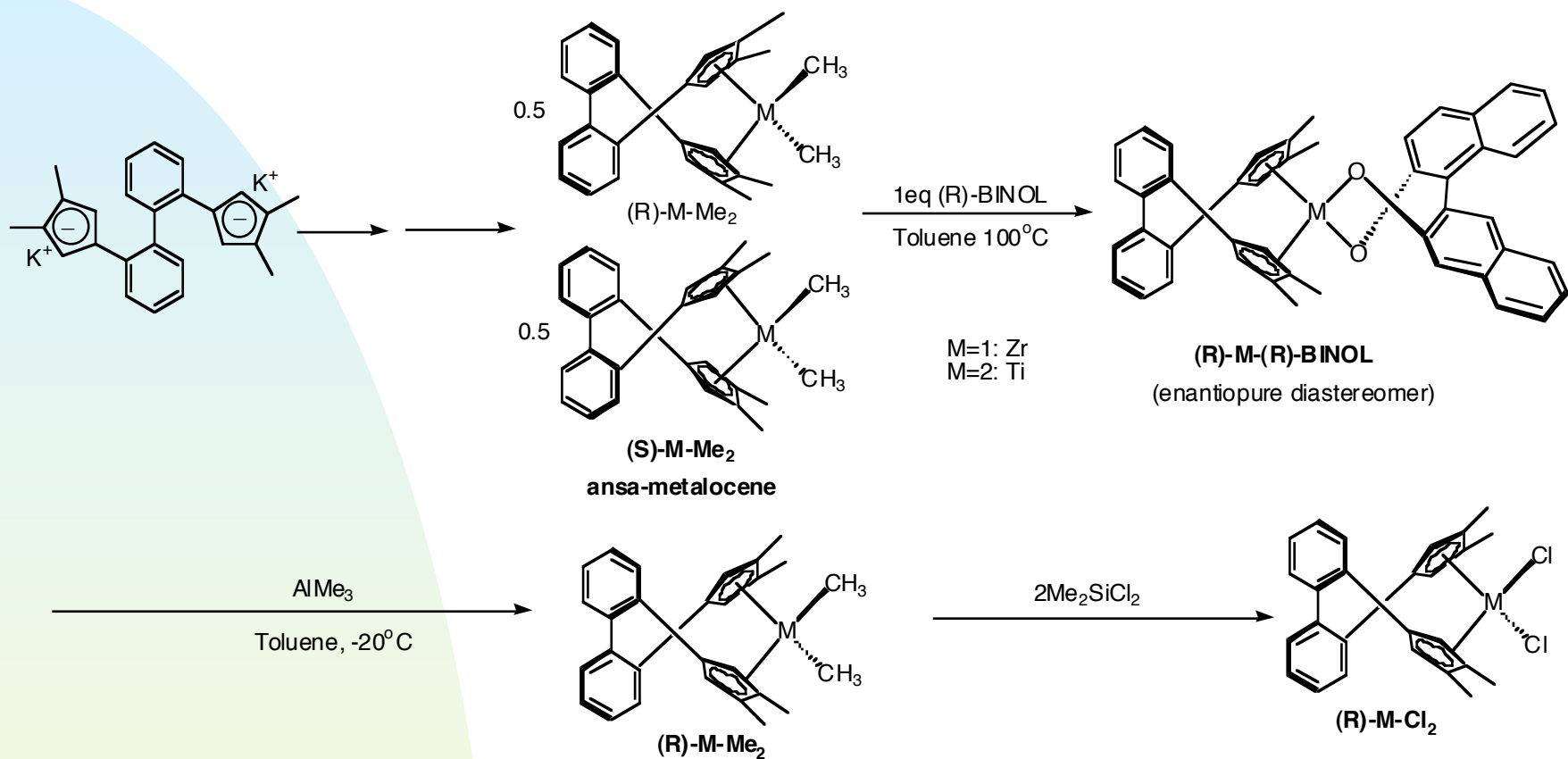


Costa, A. M.; Jimeno, C.; Gavenonis, J.; Carroll, P. J. Walsh, P. J. *J. Am. Chem. Soc.* **2002**, 124, 6929-6941

# Chiral Relay in Catalyst



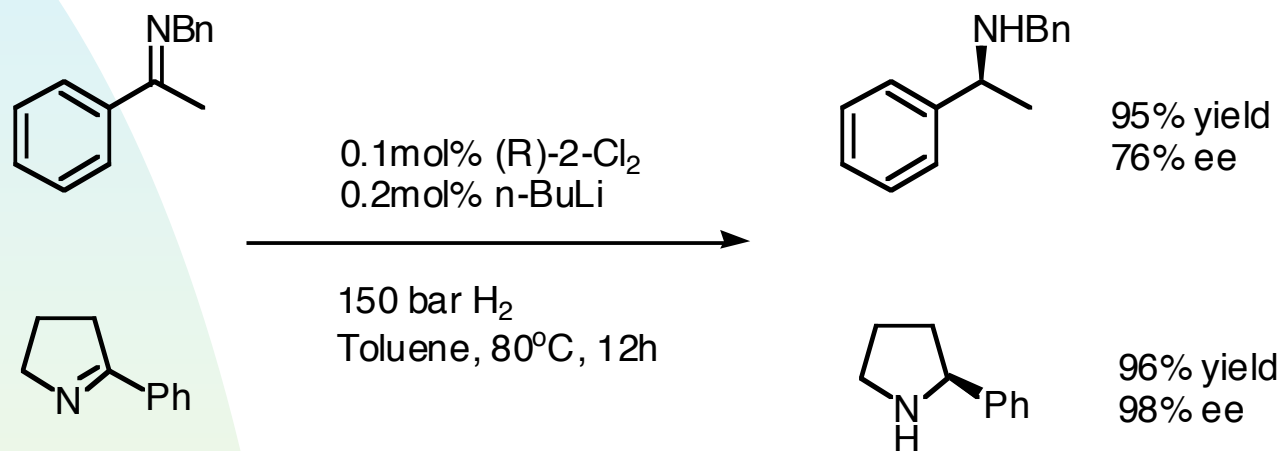
# Chiral Relay in Catalyst



Huttenloch, M. E.; Dorer, B.; Rief, U.; Prosenc, M-H.; Schmidt, K.; Bringtzing, H. H.; *J. Organomet. Chem.* **1997**, 541, 219

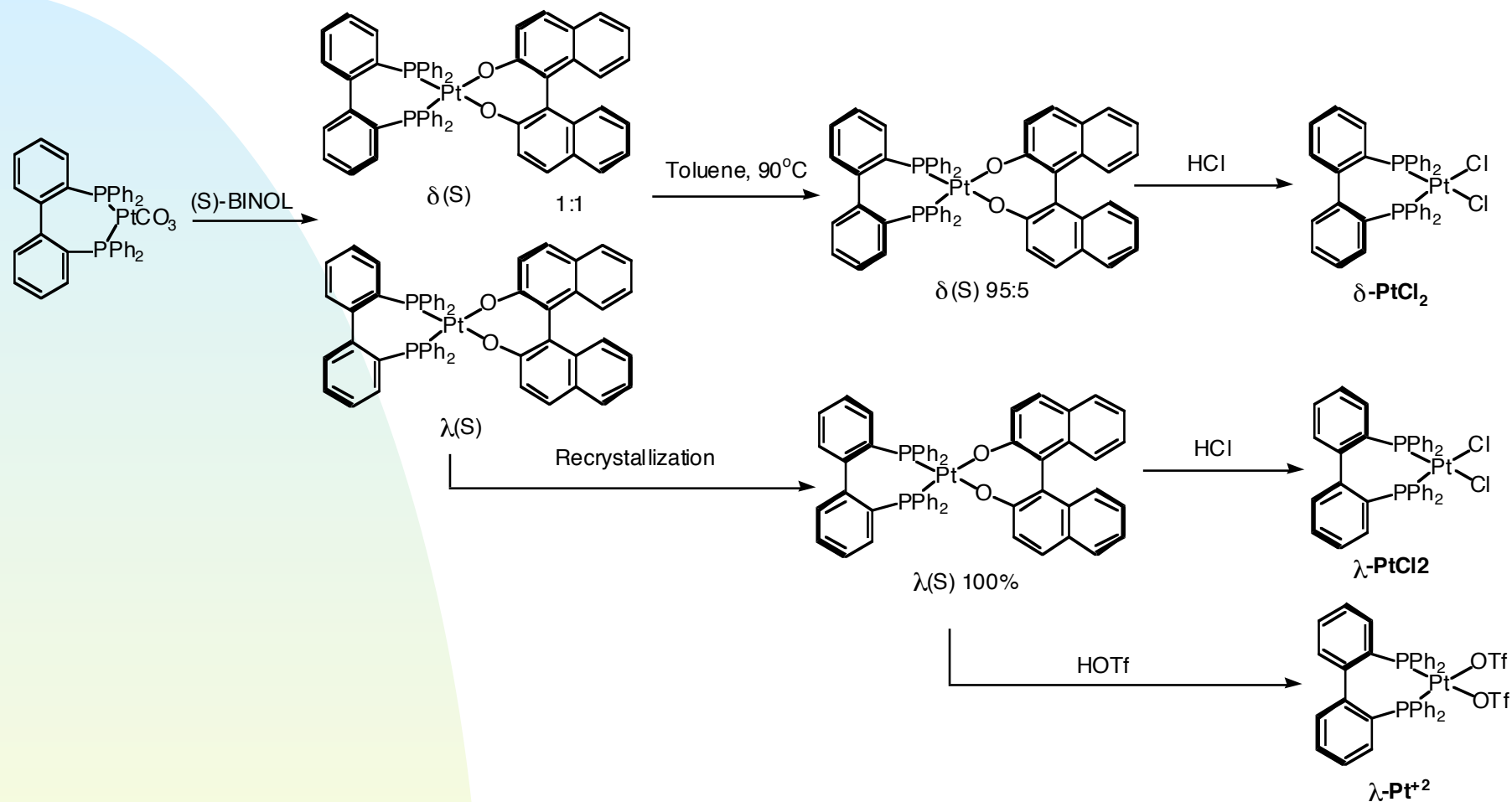
Ringwald, M.; Sturmer, R.; Brintzinger, H. H. *J. Am. Chem. Soc.* **1999**, 121, 1524-1527

# Chiral Relay in Catalyst



Ringwald, M.; Sturmer, R.; Brintzinger, H. H. *J. Am. Chem. Soc.* **1999**, 121, 1524-1527

# Chiral Relay in Catalyst

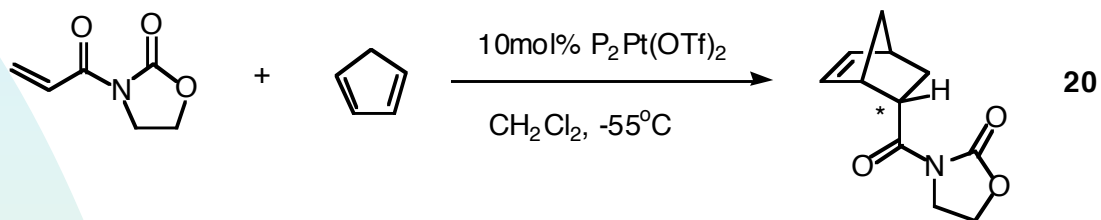


\* ( $\lambda/\delta$ ) are arbitrarily assigned to the biphep stereochemistry in terms of its skew conformation

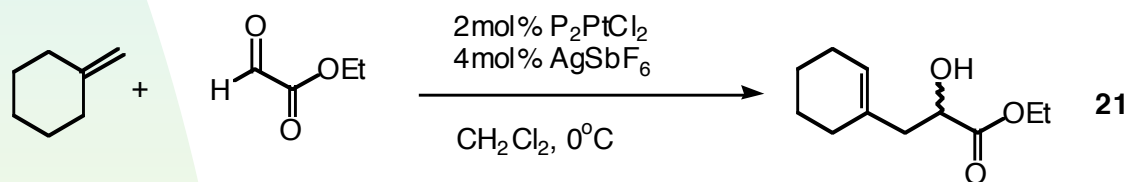
Tudor, M. D.; Becker, J. J.; White, P. S.; Gagne, M. R. *Organometallics* **2000**, 19, 4376

Becker, J. J.; White, P. S.; Gagne, M. R. *J. Am. Chem. Soc.* **2001**, 123, 9478-9479

# Chiral Relay in Catalyst



$\lambda$ - $Pt^{+2}$ : **(S)-20**, 92-94% ee, 93:7 endo/exo  
 $\delta$ - $Pt^{+2}$ : **(R)-20**, 92-94% ee, 94:6 endo/exo



$\lambda$ - $PtCl_2$ : **(S)-21**, 71% ee  
 $\delta$ - $PtCl_2$ : **(R)-21**, 70% ee

Tudor, M. D.; Becker, J. J.; White, P. S.; Gagne, M. R. *Organometallics* **2000**, 19, 4376

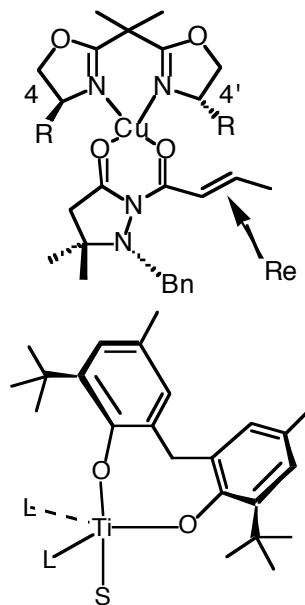
Becker, J. J.; White, P. S.; Gagne, M. R. *J. Am. Chem. Soc.* **2001**, 123, 9478-9479

## Problems in this model

(1) Straightforward experimental evidence is inadequate to support the relay mechanism. Stereochemical outcome is still not predictable in a majority of reactions.

(2) The role of relay ligand has not been well decoupled with chiral ligand. It's still hard to determine which one plays a major part in catalyst.

(3) Some other concepts describing the similar events in catalysts have been proposed, such as achiral additives,<sup>[24]</sup> ligand acceleration,<sup>[18]</sup> asymmetric activation,<sup>[25]</sup> and asymmetric poisoning.<sup>[26]</sup> It is difficult to conclude which one is closer to the real picture at least by now.



For review, Vogel, E. M.; Groger, H; Shibasaki, M. *Angew. Chem. Int. Ed.* **1999**, 38, 1570-1577

Long, J; Ding, K. *Angew. Chem. Int. Ed.* **2001**, 40, 452-457

Faller, J. W.; Laviorie, A. R.; Grimmond, B. J. *Organometallics* **2002**, 21, 1662-1666

## Future Work

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- (1) Mechanism elucidation
- (2) Experimental evidence for the relay process
- (3) Kinetic study for decoupling different pathways
- (4) Efficient design of relay template and ligand
- (5) Theoretic modeling may provide insight for relay process



## Conclusion

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1. Chiral relay is a conceptually novel idea and has been successfully applied in many reaction systems.
2. The mechanism of chiral relay is not very clear and problems still exist.
3. Future work should be done to elucidate the mechanism and direct the design of novel chiral relay groups.

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